

Phosphorus recovery process

Phosphogreen

Mathieu DELAHAYE
SUEZ - Innovation Manager

Context

Context

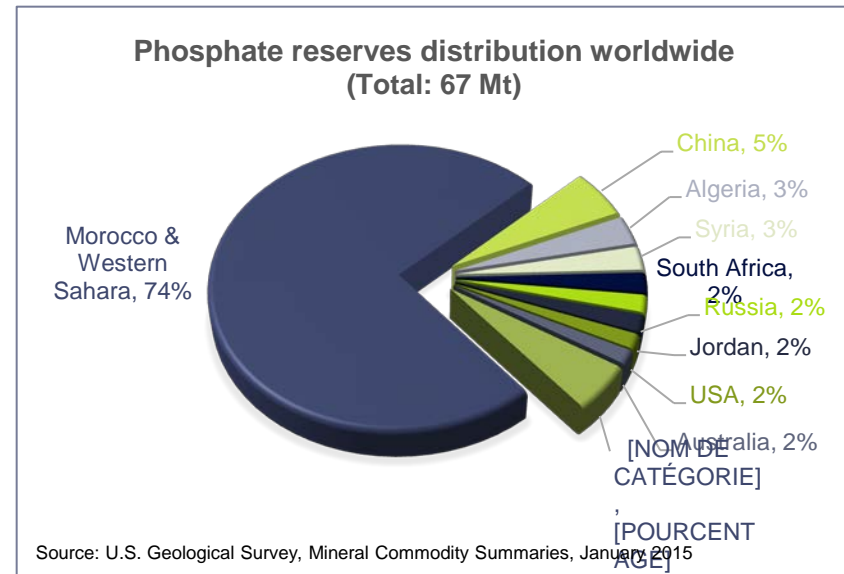
Facts and figures about phosphorus

today, a nutrient

- essential to life (creation of DNA, cell membranes...)
- non-substitutable
- intensively used as an additive in food industry and agriculture

tomorrow, a geopolitical issue?

- depletion foreseen by the end of 21st century
- reserves unequally distributed throughout the world:
95 % of the reserves located in only 9 countries
- complicated geopolitical situation in some countries
- Europe has *no* phosphate reserves



Solution

Phosphorus recovery

considering that

- 20% of current world demand of phosphorus could be covered by recovery from wastewater
- 80% of phosphorus extracted from phosphate ores is intended for fertilizers

the solution is

to convert **phosphorus** present in wastewater into a **valuable fertilizer**

to help our clients take a step forward towards **sustainable development** and
circular economy

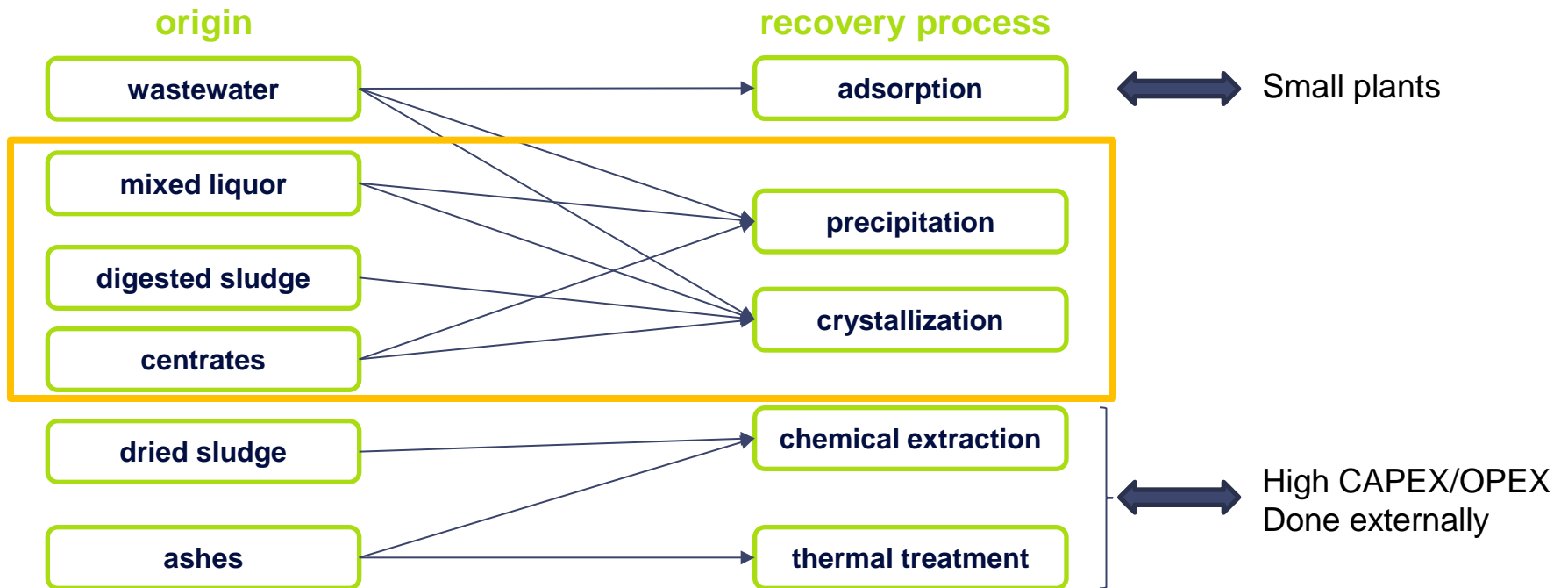
Phosphorus recovery processes

P-recovery processes

one aim

extract phosphorus from wastewater in a form that can be used directly as a fertilizer or as a raw material for industries.

various processes



P-recovery processes

Precipitation-crystallization reactions

Dissolved phosphate ions can be converted into solid phase by precipitation-crystallization.

	possible recovery of phosphorus?	use of the product as fertilizer?
<ul style="list-style-type: none"> $\text{PO}_4^{3-} + \text{Fe}^{3+} \rightarrow \text{FePO}_4$ ferric chloride, FeCl_3 	no	no
<ul style="list-style-type: none"> $\text{PO}_4^{3-} + \text{Al}^{3+} \rightarrow \text{AlPO}_4$ aluminum sulphate, $\text{Al}_2(\text{SO}_4)_3$ 	yes (electrothermal treatment)	no
<ul style="list-style-type: none"> $2 \text{PO}_4^{3-} + 3 \text{Ca}^{2+} \rightarrow \text{Ca}_3(\text{PO}_4)_2 - \text{HAP}$ lime, $\text{Ca}(\text{OH})_2$ 	yes (electrothermal treatment)	low bioavailability
<ul style="list-style-type: none"> $\text{PO}_4^{3-} + \text{Mg}^{2+} + \text{NH}_4^+ \rightarrow \text{MgNH}_4\text{PO}_4 \cdot 6 \text{H}_2\text{O} - \text{MAP}$ magnesium chloride, MgCl_2 <p style="text-align: center;"> $\text{MgNH}_4\text{PO}_4 \cdot 6 \text{H}_2\text{O}$ struvite </p>	yes (as is)	yes

Phosphogreen Technology description

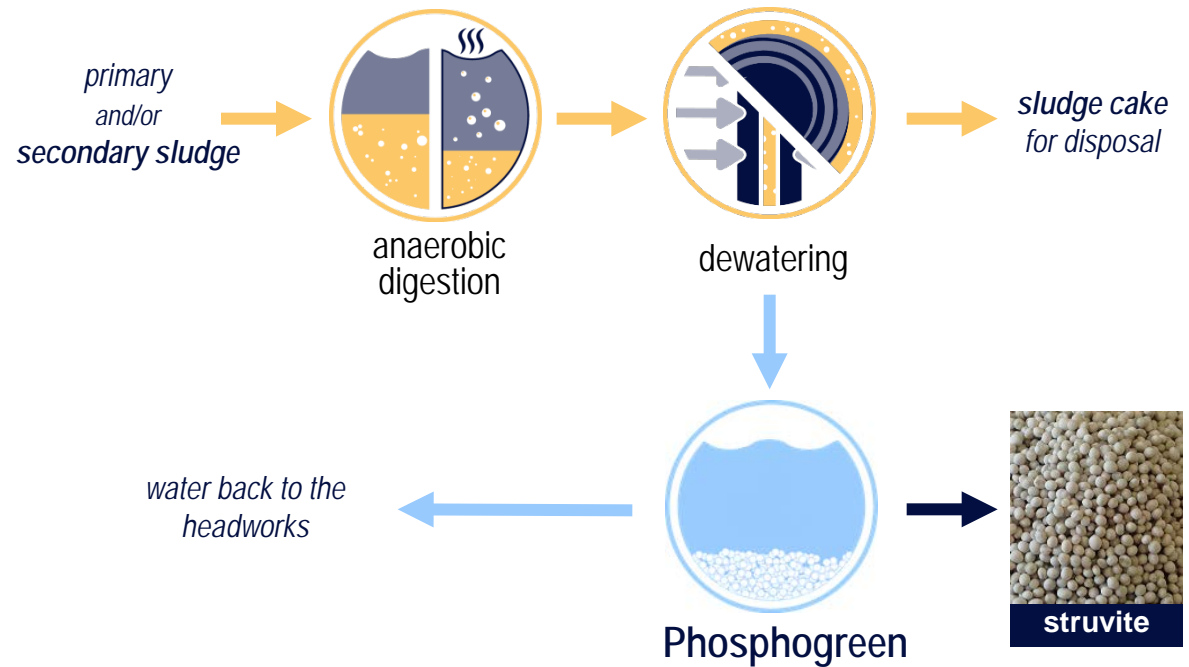
Phosphogreen

Principle

application field

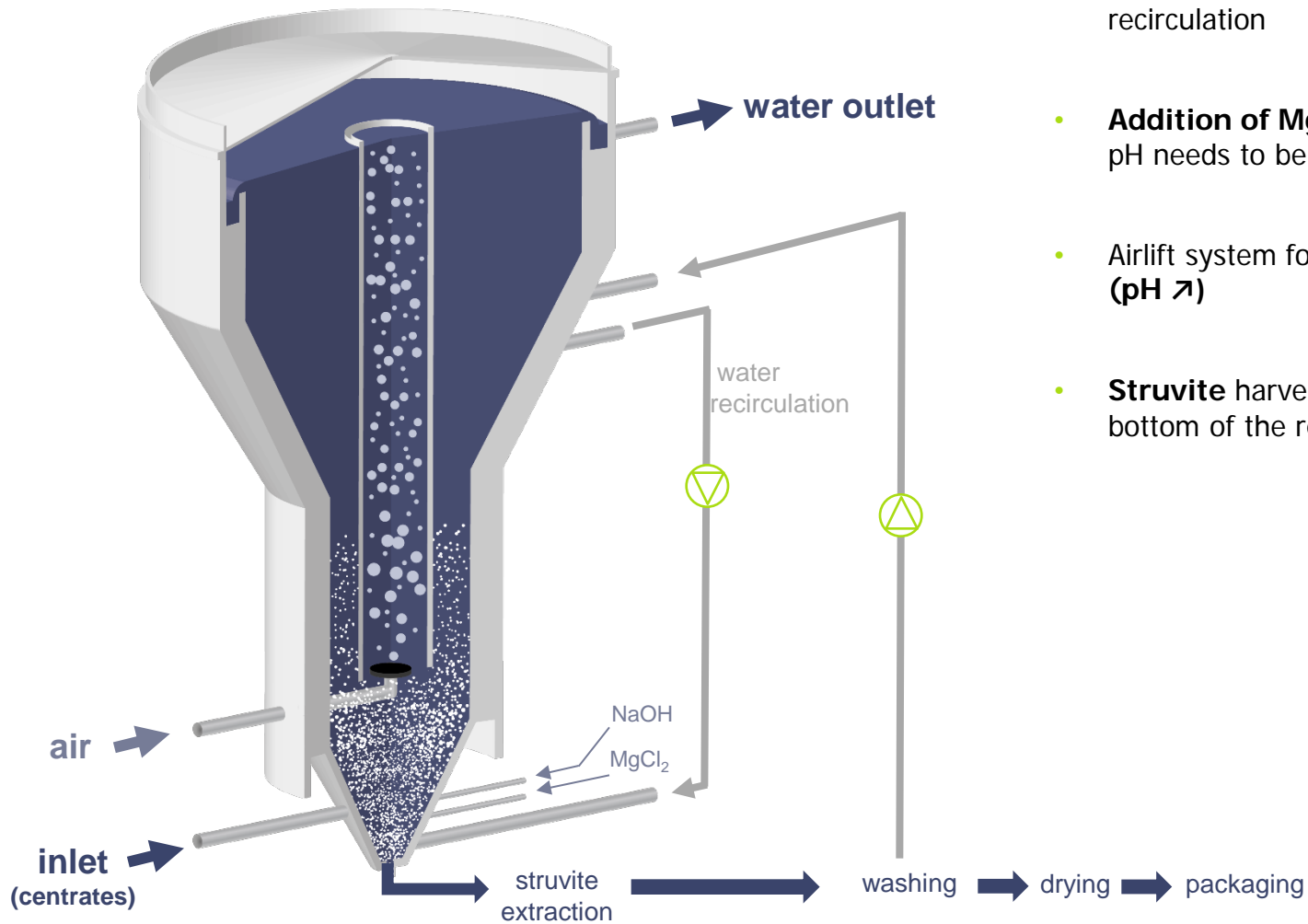
- WWTP capacity > 40,000 PE
- Bio-P removal
- Anaerobic digestion
- Phosphorus in digested sludge concentrates ≥ 70 mg/L

position in the treatment line



Phosphogreen

Process overview



- **Fluidized bed reactor** with recirculation
- **Addition of $MgCl_2$ and $NaOH$** (if pH needs to be adjusted)
- Airlift system for **CO_2 stripping** (pH \nearrow)
- **Struvite** harvested from the bottom of the reactor

Phosphogreen

Struvite characteristics

Struvite ↔ $\text{MgNH}_4\text{PO}_4 \cdot 6 \text{H}_2\text{O}$

- granule size: 1 to 3 mm
 - including
 - 12.2% w/w phosphorus
 - 5.8% w/w nitrogen
 - 10.4% w/w magnesium
 - heavy metals content (mg/kg):

	struvite	Danish standards for sludge
Cd	0.06	0.8
Hg	<0.1	0.8
Pb	<0.2	120
Ni	0.5	30
Cr	4.5	100
Zn	7.4	4000
Cu	0.8	1000



References

Phosphogreen references

Åby (Aarhus, Denmark)

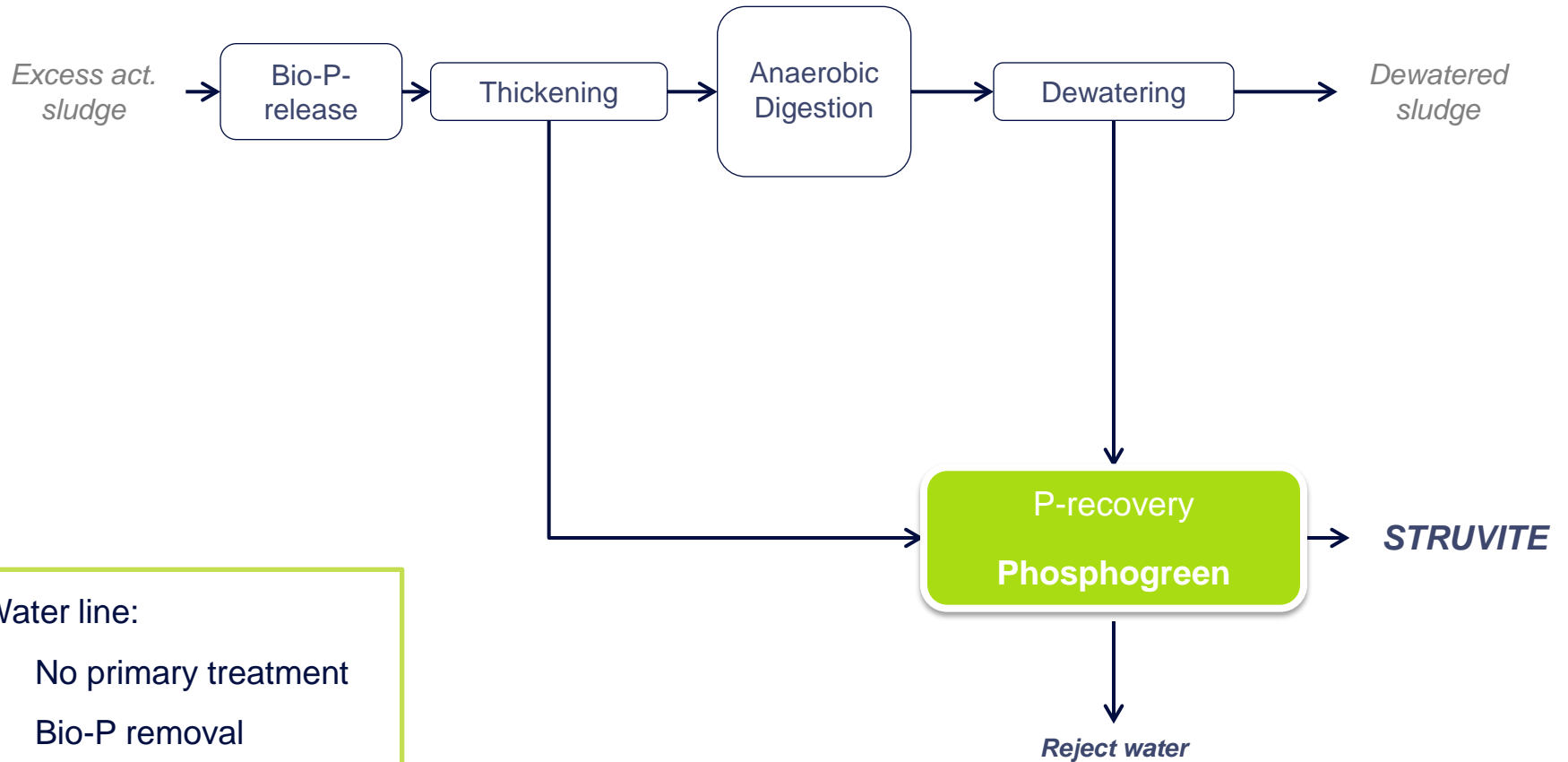
plant's characteristics

- 84,000 PE
 - No primary treatment
 - P-recov. unit start-up date: 2013
 - **WWTP's inlet:**
 - 450 kg/d N_{tot}
 - 105 kg/d P_{tot}
 - **Outlet:**
 - 300 kg/d struvite
 - (incl. 37 kg P/d)
- ➔ ~ 35% of P recovered



Phosphogreen references

Åby (Aarhus, Denmark)



Water line:

- No primary treatment
- Bio-P removal

Phosphogreen references

Herning (Denmark)

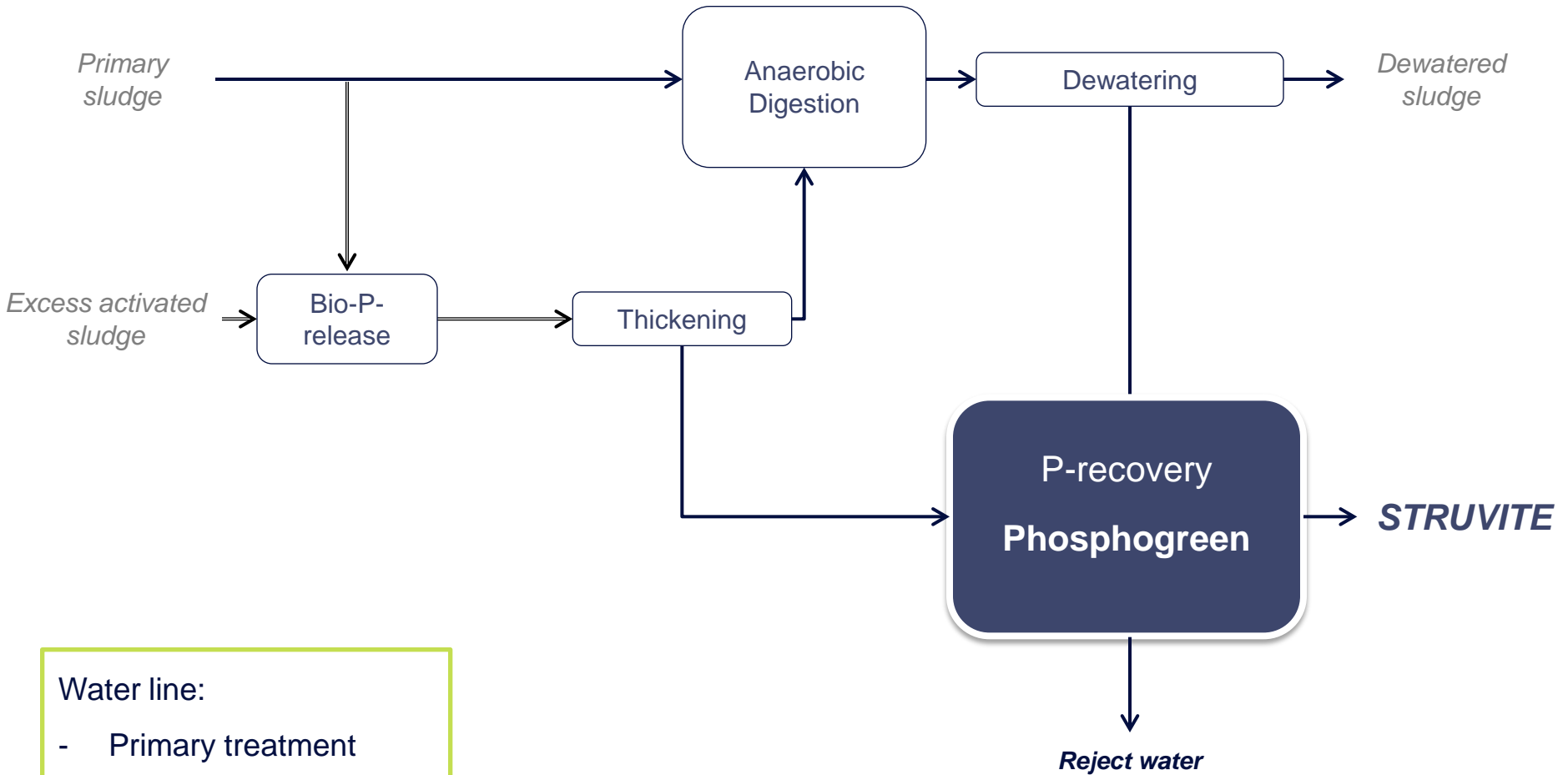
plant's characteristics

- 150,000 PE
 - Primary treatment
 - P-recov. unit start-up date: 2015
 - **WWTP's inlet:**
1200 kg/d N_{tot}
240 kg/d P_{tot}
 - **Outlet:**
290 kg/d struvite
(incl. 36 kg P/d)
- ➡ ~ 15 % of P recovered



Phosphogreen references

Herning (Denmark)



Water line:

- Primary treatment
- Bio-P removal

Benefits

Phosphogreen

Benefits

○ financial revenue

- Sale of struvite as fertilizer (75 to 300 €/t)
- ROI 5 to 10 years

○ savings on various levels of the process

- **reduction of chemicals consumption** for phosphorus physicochemical treatment
- **less energy** consumed for biological treatment (removal of a part of nitrogen by struvite precipitation)
- **increased equipment service life** thanks to controlled and localized precipitation of struvite
- **reduction of sludge disposal costs** by minimization of sludge volumes

○ commitment to sustainable development

- **resource recovery** (phosphorus, nitrogen): contribution to the circular economy
- **lower environmental footprint** through energy savings, sludge volume reduction, increased equipment service life, reduced chemical consumption

Regulation constraints

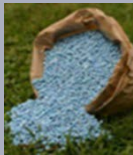
European legislative and regulatory framework

the regulatory framework is complex for phosphorus recovery projects:



spreading plan

- limited quantities
- many quality controls



fertilizer regulations

- national standards
- EU directive on fertilizers



national approval
(usually necessary)

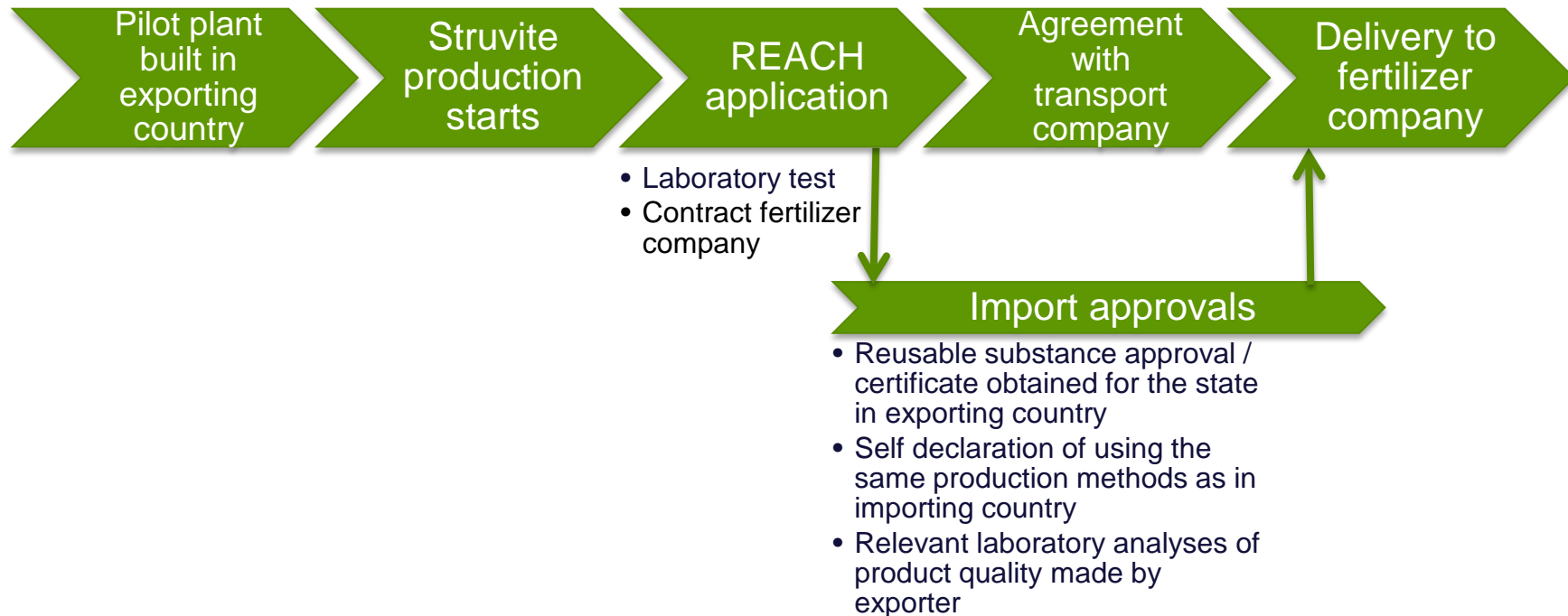
A significant number of laws and regulations apply in Europe, which complicates the phosphorus recycling projects. Simplification's works are underway (end of waste status).

Regulation constraints

○ Struvite homologated as a fertilizer in some countries (DK, UK), but not in others (FR, ...)

- No standardized European regulation
- Slow down the development of P recycling plants

○ Difficulties to import/export struvite between EC members while there is a market



Thank you for your attention